of the determination of the C-terminal sequence of sturine B [1] enabled the complete amino-acid sequence of sturine B to be established: H-Ala-Arg₅-Ser-Ser-Arg-Pro-Glx-Arg₆-His-Gly-Arg₂-Gly-Arg₄-OH.

LITERATURE CITED

- 1. L. K. Evseenko, E. P. Yulikova, and A. B. Silaev, Khim. Prirodn. Soedin., 778 (1975) [in this issue].
- 2. F. Sanger, Biochem. J., 39, 507 (1945).
- 3. W. Gray, Methods in Enzymology, 11, 139 (1968).

SYNTHESIS OF A HEXAPEPTIDES RELATED TO ELEDOISIN

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UDC 547.964.4

In order to investigate the mechanism of the action of kinins, we have synthesized by a new method lysylphenylalanylisoleucylglycylleucylmethionine amide (I, 1-6, Table 1, Scheme) — a substance possessing a strong hypotensive and myotropic action [1]. In all cases (with the exception of E 5-6) for the formation of the peptide bond we used 2-ethoxy-1-ethoxycarbonyl-1,2-dihydroquinoline (EEDQ) [2] in solution in diformamide or ethanol (for H 1-6) in the presence of N-methylmorpholine. Compound E 5-6 was obtained as described by Lübke et al. [3]. After the transfer of the reaction products into ethyl acetate, the starting materials were separated by washing the solutions with sodium bicarbonate and potassium bisulfate.

The benzyloxycarbonyl and tert-butyloxycarbonyl groups were split off by treatment with solutions of hydrogen bromide and hydrogen chloride in acetic acid, respectively; the p-nitrobenzyl ester was cleaved by catalytic hydrogenolysis.

Compound E 2-4 was recrystallized from water, G 1-4 from 50% methanol, and H 1-6 from a mixture of dimethylformamide and water. The final product I 1-6 was purified by dissolution in water, filtration, and lyophilization of the filtrate; mp 241°C (decomp.) (240-243°C, decomp. [4]), $[\alpha]_D^{23}$ -16.3° (c 0.8; acetic acid) (-16.8° [4]). The results of amino-acid and elementary analyses agreed with the calculated figures. In the biological test* (contraction of the guinea-pig ileum), I 1-6 showed an activity more than twice as great as that of bradykinin, which agrees with literature information [4]. (See scheme.)

TABLE 1. Electrophoretic and Chromatographic Constants of the Compounds Synthesized

Peptide	E _{His} *	Rf in the following systems				
		1	2	3	4	5
B 3-4 C 3-4	0,71	0,45	0,95 0,10	0,98 0,50	0,85 0,50	0,96 0,50
D 2-4 E 2-4	0 0,56	0,40	0,95 0,35	0,98 0,90	0,90	0,85 0,80
F 1-4	0	0,18	0,95 0,90	0,90 0,97	0,85 0,86	0,96 0,88
H 1-6	0 0,76	0,02	0,90 0	0,90 0,70	0,86 0,55	0,88 0,20

*The electrophoretic mobility E_{His} was determined on type "S" [medium] paper (Leningrad Paper Mill No.2) in 1 N acetic acid.

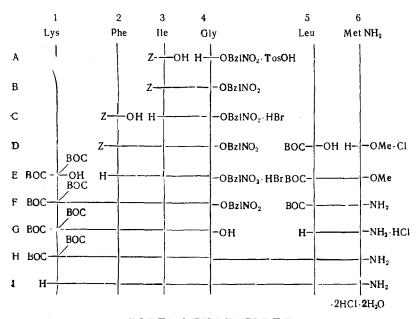
†For thin-layer chromatography on silica gel we used "Silufol" plates and the following solvent system: 1) chloroform—acetic acid (95:5); 2) chloroform—methanol—acetic acid (85:10:5); 3) n-butanol—acetic acid—pyridine—water (15:3:10:12); 4) n-propanol—concentrated aqueous ammonia (84:37); 5) n-butanol—isopropanol—water—mono-chloroacetic acid (65 ml:15 ml:20 ml:3 g).

Institute of Organic Synthesis, Academy of Sciences of the Latvian SSR. Translated from Khimiya Prirodnykh Soedinenii, No. 6, pp. 818-820, November-December, 1975. Original article submitted May 2!

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^{*}The tests were performed by Z. P. Auna and V. E. Klusha.

Scheme of the Synthesis



LITERATURE CITED

- 1. L. Bernardi, S. Bosisio, F. Chillemi, G. de Caro, R. de Castigesone, V. Erspamer, A. Glasser, and O. Goffredo, Experientia 20, 306 (1964).
- 2. B. Belleau, R. Martel, G. Lacasse, M. Menard, N. L. Weinberg, and Y. G. Perron, J. Amer. Chem. Soc., 90, 823 (1968); L. Fieser and M. Fieser, Reagents for Organic Synthesis, Wiley, New York [Russian translation, Moscow, Vol. 5 (1971), p. 236].
- 3. K. Libke, E. Schröder, R. Schmiechen, and H. Gibian, Ann. Chem., 679, 195 (1964).
- 4. S. Sakakibara and M. Fujino, Bull. Chem. Soc. Japan, 39, 947 (1966).

LOCALIZATION OF THE ACTIVITY OF LEUCINE
AMINOPEPTIDASE AFTER ELECTROPHORESIS IN
ACRYLAMIDE GEL

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UDC 577.156-537.363

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A method has been developed for localizing the activity of leucine aminopeptidase after electrophoresis in acrylamide gel with the aid of the substrate L-leucine p-nitroanilide [2].

The leucine aminopeptidase was isolated from "orizin" – a mixture of the proteins of Asp. oryzae (obtained in the Moscow enzyme factory) by precipitation with ethanol and chromatography on columns of DEAE-cellulose at pH 5.6 and 6.9 and hydroxylapatite. Disk electrophoresis was performed at pH 8.3 in tris-glycine buffer on a "Canalco-Europe" instrument (Holland), using an 11.2% acrylamide gel and 70×6 mm tubes. The duration of an experiment was 1.5 h at a current strength of 3 mA/gel and a temperature

All-Union Scientific-Research Institute of the Genetics and Breeding of Industrial Microorganisms. Translated from Khimiya Prirodnykh Soedinenii, No. 6, p. 820, November-December, 1975. Original article submitted June 4, 1975.

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